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COURSE SPECIFICATIONS

Pharmaceutical Organic ChemistryIII

Second Level– Semester 3 2020-2021

Course specification of Pharmaceutical Organic Chemistry-3

University: Zagazig

Faculty: Pharmacy

A- Course specifications:

Program on which the course is given: Bachelor of Pharmacy (Pharm D) Major or Minor element of programs: Major Department offering the program: ------Department offering the course: Pharmaceutical Organic Chemistry Academic year Level: level two / semester 3 Date of specification approval: Oct 2020

B- Basic information:

Title: Code: PR 303

Credit Hours: ---

Lectures: 2hr/week

Practical: 1hr/week

Tutorials: ---

Total: 3hrs/week

C- Professional information:

1-Overall aim of the course:-

On completion of the course, students will be able to:

- Outline the chemistry of carbohydrates, their identification, and their application in pharmaceutical compounds.
- Identify the chemistry of heterocyclic compounds.
- Explain the principles of spectroscopy ((UV, IR, ¹ HNMR, ¹³CNMR & mass spectroscopy) and how to interpret the spectroscopic data of different compounds.

2- Key elements of pharmaceutical organic chemistry 3.

DOMAIN 1- FUNDAMENTAL KNOWLEDGE

1-1- COMPETENCY: Integrate knowledge from basic and applied pharmaceutical and clinical sciences to standardize materials, formulate and manufacture products, and deliver population and patient-centered care.

1.C1.1.	Outline the chemistry of carbohydrates and their important chemical reactions.										
1.C1.2.	• Illustrate the principles of spectroscopic techniques (UV, IR, NMR and Mass spectroscopy) in identification of aromatic compounds.										
1.C1.3.	 Identify chemical name and different synthetic pathways for pharmaceutical heterocyclic compounds including commercially available drugs. 										
DOMAI	N 2: PROFESSIONAL AND ETHICAL PRACTICE										
pharmace	APETENCY: Standardize pharmaceutical materials, formulate and manufacture utical products, and participate in systems for dispensing, storage, and on of medicines.										
2.C2.1.	 Synthesize different pharmaceutically active nuclei including pyrazole, benzotriazole, benzothiophene and quinoxalinone 										
2.C2.2.	Perform identification tests for carbohydrates.										
2.C2.3	• Perform Purification of different target compounds using the separation techniques.										
2.C2.4.	• select the proper spectroscopic technique for pharmaceutical compounds identification.										
	IPETENCY: Handle and dispose biologicals and synthetic/natural utical materials /products effectively and safely with respect to relevant laws and ns.										
2.C3.1	Handle basic laboratory equipment and chemicals effectively and safely.										
2.C3.2	Adopt GLP guidelines for safe handling of chemicals										
DOMAI	N 4: PERSONAL PRACTICE										
	IPETENCY : Express leadership, time management, critical thinking, problem										
4.C1.1.	 ndependent and team working, creativity and entrepreneurial skills. Work effectively as a member of a team. 										
4.C1.2.	Demonstrate problem solving skills.										

4-2- COMPETENCY: Effectively communicate verbally, non-verbally and in writing
with individuals and communities.
4.C2.1. • Communicate effectively with others.

D- Contents:

Week	Lecture contents (2hr/week)	Practical session (1hr/week)
No.		
1	 I- Carbohydrates Introduction and Classification. Monosaccharides i. Synthesis ii.Cyclic structure of monosaccharides iii.Chemical reactions of monosaccharides activity 	Lab. Safety
2	*Disaccharides Nomenclatureand determination of the type of glycosidic linkage. *.Polysaccharides Nomenclature and determination of the type of glycosidic linkage. *Drugs containing carbohydrates	Identification of carbohydrates Identification of glucose and fructose
3	II- Heterocyclic chemistry Nomenclature of heterocyclic compounds	Identification of carbohydrates Identification of sucrose, lactose and starch
4	Five-membered heterocyclic with one heteroatom, synthesis and reactions	Synthesis of fructosasone
5	 Indole (benzo[b]pyrrole) Five-membered rings containing two heteroatoms. synthesis and reactions 	Synthesis of β-penta-acetyl glucose
6	Six-membered heterocyclic with one heteroatom: synthesis, reactions and applications.	Synthesis of 3,5-dimethyl pyrazole
7	Mid. term exam	Mid. term exam
8	Quinoline and isoquinoline, synthesis and reactions	Synthesis of 5-nitrosalicylic acid
9	Six-memberedringscontainingtwoheteroatoms+purinenucleusandapplication. </th <th>Synthesis of 3-methyl-1H-quinoxalin- 2-one</th>	Synthesis of 3-methyl-1H-quinoxalin- 2-one
10	III- Spectroscopy UV and visible spectroscopy	Synthesis of 1,2,3-benzotriazole
11	Infrared spectroscopy Applications on infrared spectroscopy	Synthesis of ethyl 2-amino-4,5,6,7- tetrahydrobenzo(b)thiophene-3- carboxylate
12	Mass spectrometry	Activity (Synthesis of certain drugs containing heterocycles)
13	¹ H-NMR spectroscopy	Practical exam

14	¹³ C-NMR spectroscopy	Practical exam
15	Final written exam	

E- Teaching and Learning Methods:

- Lectures
- Practical session
- Group discussion
- Blended learning

F- Student Assessment methods:

- Written exams to assess:1.C1.1. ,1.C1.2, 1.C1.3., 4.C2.1
- Practical exams to assess: 2.C2.1, 2.C2.2, 2.C2.3, 2.C2.4, 2.C3.1, 2.C3.2
- Oral exam **to assess:** 1.C1.1. ,1.C1.2,1.C1.3, 4.C2.1
- Activities within labs to assess: 4.C1.1, 4.C1.2, 4.C2.1

Assessment schedule

Assessment (1): Written exams	Week 15
Assessment (2): Practical exams	Week 13, 14
Assessment (3): Oral exams	Week 15
Assessment (4): Activity	participation each lab
Assessment (5): Periodical exams	Week 7

Weighting of Assessment

Assessment method	Marks	Percentage
Periodicexam	10	10%
Final written exam	50	50%
Activity	5	5%
Practical practice & exam	25	25%
Oral exam	10	10%
TOTAL	100	100%

G- Facilities required for teaching and learning:

• For lectures: White boards, data show.

For Labs: Chemicals, glassware, instruments, Digital balances, water

bathes

H- List of References:

1- Course Notes:

- Presentation Printout.

- Student book of Pharmaceutical Organic Chemistry approved by the department

2- Essential Books:

- ✓ ChaudenJaidey, 2018, Organic Spectroscopy; Delhy.
- ✓ Francis A. Carey, 2009, Organic Chemistry; 9th Edition, McGraw-Hill
- ✓ T. W. Graham Solomons and Craig B. Fryhle, 2010, Organic

Chemistry; 11th Edition, John Willy & Sons Inc, USA.

Course Coordinator: Prof. Dr. Eatedal Abdel-aal

Head of Department: Prof. Dr. Hanan Abdelrazik Adelfattah

تم مناقشة و إعتماد توصيف المقرر من مجلس القسم بتاريخ /

		Matrix I of Pharmaceutical organic Chemistry-3												
	Course Contents		Key elements of Pharmaceutical organic Chemistry-3											
			1-1- COMPETENCY		2-2- COMPETENCY				2-3- ETENCY		4-1- COMPETENCY		- ГENC	
	Lectures		1 (1)	1 (1 2								Y		
			1.C1.2	1.C1.3	2.C2.1	2.C2.2	2.C2.3	2.C2.4	2.C3. 1	2.C3. 2	4.C1.1	4.C1.2	4.C2.	1
1	 Introduction and Classification. Monosaccharides i. Synthesis ii.Cyclic structure of monosaccharides iii.Chemical reactions of monosaccharides activity 	X												
2	Disaccharides Nomenclatureand determination of the type of glycosidic linkage. *.Polysaccharides Nomenclature and determination of the type of glycosidic linkage. *Drugs containing carbohydrates	x												
3	II- Heterocyclic chemistry Nomenclature of heterocyclic compounds			X										
4	Five-membered heterocyclic with one heteroatom, synthesis and reactions			X										
5	 Indole (benzo[b]pyrrole) Five-membered rings containing two heteroatoms. synthesis and reactions 			X										
7	Six-membered heterocyclic with one heteroatom: synthesis, reactionsand applications.x			X										

			-			-		-			
Quinoline and isoquinoline, synthesis and		Χ									
Six-membered rings containing two		v									
		Λ									
	Χ										
· · · · · ·											
	Χ										
	X										
¹³ C-NMR spectroscopy	Х										
						_					
Lab. Safety							Χ	X		X	
Identification of Carbohydrates				X	Χ	Х	V	V		V	
Identification of glucose and fructose							Λ	Λ		Λ	
Identification of Carbohydrates				X	Χ	Х					
							Х	Х		Χ	
starch											
Synthesis of fructosazone			Χ		X	Х	Х	Х		X	
Synthesis of β-penta-acetyl glucose			Χ		Χ	Х	Х	Х		Χ	
Synthesis of 3,5-dimethyl pyrazole			Χ		Χ	Χ	X	Х		Χ	
Synthesis of 5-nitrosalicylic acid			Χ		X	Х	Х	Х		Χ	
Synthesis of 3-methyl-1H-quinoxalin-			v		X	Х	v	v		V	
2-one			Δ				Λ	Λ		Δ	
Synthesis of 1,2,3-benzotriazole			Χ		Χ	Х	Х	X		Χ	
Synthesis of ethyl 2-amino-4,5,6,7-			X		X	Х	Х	X		X	
	reactionsSix-membered rings containing two heteroatoms + purine nucleus and application.INFrame spectroscopyUV and visible spectroscopyUV and visible spectroscopyInfrared spectroscopyApplications on infrared spectroscopyMass spectroscopyMass spectroscopy ¹ H-NMR spectroscopy ¹³ C-NMR spectroscopyIdentification of Carbohydrates Identification of Carbohydrates Identification of Sucrose, lactose and starchSynthesis of fructosazoneSynthesis of fructosazoneSynthesis of S-nitrosalicylic acidSynthesis of 3-methyl-1H-quinoxalin- 2-oneSynthesis of 1,2,3-benzotriazole	reactionsImage: containing two heteroatoms + purine nucleus and application.III- SpectroscopyXUV and visible spectroscopyXInfrared spectroscopyXApplications on infrared spectroscopyXMass spectrometryX ¹ H-NMR spectroscopyX ¹³ C-NMR spectroscopyXLab. SafetyXIdentification of Carbohydrates1Identification of Sucrose, lactose and fructose1Identification of sucrose, lactose and starch1Synthesis of β-penta-acetyl glucose5Synthesis of 3,5-dimethyl pyrazole5Synthesis of 3-methyl-1H-quinoxalin-2-one1Synthesis of 1,2,3-benzotriazole1	reactionsImage: containing two heteroatoms + purine nucleus and application.XIII- SpectroscopyXUV and visible spectroscopyXUV and visible spectroscopyXApplications on infrared spectroscopyXMass spectrometryXIH-NMR spectroscopyXI'H-NMR spectroscopyXI'aC-NMR spectroscopyXI'aC-NMR spectroscopyXI'achon of CarbohydratesIIdentification of CarbohydratesIIdentification of Sucrose, lactose and starchISynthesis of fructosazoneISynthesis of 3,5-dimethyl pyrazoleISynthesis of 3-methyl-1H-quinoxalin- 2-oneISynthesis of 1,2,3-benzotriazoleISynthesis of 1,2,3-benzotriazoleIIISynthesis of 1,2,3-benzotriazoleIIISynthesis of 1,2,3-benzotriazoleIII	reactionsAASix-membered rings containing two heteroatoms + purine nucleus and application.XXIII- SpectroscopyXXUV and visible spectroscopyXXInfrared spectroscopyXXApplications on infrared spectroscopyXIMass spectrometryXI ¹¹ H-NMR spectroscopyXI ¹³ C-NMR spectroscopyXIIdentification of Carbohydrates Identification of glucose and fructoseIIIdentification of Carbohydrates Identification of Sucrose, lactose and starchIISynthesis of β-penta-acetyl glucoseIXXSynthesis of 3,5-dimethyl pyrazoleIIXSynthesis of 3-methyl-1IH-quinoxalin- 2-oneIIXSynthesis of 1,2,3-benzotriazoleIIX	reactionsAASix-membered rings containing two heteroatoms + purine nucleus and application.XXIII- Spectroscopy UV and visible spectroscopyXXIII- Spectroscopy UV and visible spectroscopyXXInfrared spectroscopy Applications on infrared spectroscopyXImage: Constraint of the spectroscopyMass spectrometryXXImage: Constraint of the spectroscopyInfrared spectroscopyXImage: Constraint of the spectroscopyXImage: Image: Im	reactionsΛΛΛSix-membered rings containing two heteroatoms + purine nucleus and application.XXIII- SpectroscopyX X UV and visible spectroscopyX X Infrared spectroscopyX X Mass spectrometryX X ''H-NMR spectroscopyX X ''H-NMR spectroscopyXX''H-NMR spectroscopyXX''H-NMR spectroscopyXX <td>reactionsNNNNNSix-membered rings containing two heteroatoms + purine nucleus and application.XXNNIII- Spectroscopy UV and visible spectroscopyXXNNNInfrared spectroscopy Applications on infrared spectroscopyXNNNNMass spectrometryXXIIII¹¹H-NMR spectroscopyXXIIII11C-NMR spectroscopyXIIIIIIdentification of Carbohydrates Identification of Carbohydrates Identification of Sucrose, lactose and starchIIIXXXSynthesis of fluctosazoneIIIXXXXSynthesis of 3,5-dimethyl pyrazoleIIIXXXXXSynthesis of 1,2,3-benzotriazoleIIIXXXXXSynthesis of 1,2,3-benzotriazoleIIIXXXXX</td> <td>reactionsImage: Marking two heteroatoms + purine nucleus and application.MMImage: Marking two heteroatoms + purine nucleus and application.III- SpectroscopyXXXXXUV and visible spectroscopyXXXXXInfrared spectroscopyXXXXXAmass spectrometryXXXXXXMass spectroscopyXXXXXXMass spectroscopyXXXXXXIdentification of CarbohydratesXXXXXIdentification of sucrose, lactose and starchXXXXXSynthesis of 3.5-dimethyl pyrazoleXXXXXSynthesis of 5-nitrosalicylic acidXXXXXSynthesis of</td> <td>$\begin{array}{c c c c c c c } \hline \mathbf{reactions} & \mathbf{N} &$</td> <td>reactionsAAABACCC</td> <td>reactionsII</td>	reactionsNNNNNSix-membered rings containing two heteroatoms + purine nucleus and application.XXNNIII- Spectroscopy UV and visible spectroscopyXXNNNInfrared spectroscopy Applications on infrared spectroscopyXNNNNMass spectrometryXXIIII ¹¹ H-NMR spectroscopyXXIIII 11 C-NMR spectroscopyXIIIIIIdentification of Carbohydrates Identification of Carbohydrates Identification of Sucrose, lactose and starchIIIXXXSynthesis of fluctosazoneIIIXXXXSynthesis of 3,5-dimethyl pyrazoleIIIXXXXXSynthesis of 1,2,3-benzotriazoleIIIXXXXXSynthesis of 1,2,3-benzotriazoleIIIXXXXX	reactionsImage: Marking two heteroatoms + purine nucleus and application.MMImage: Marking two heteroatoms + purine nucleus and application.III- SpectroscopyXXXXXUV and visible spectroscopyXXXXXInfrared spectroscopyXXXXXAmass spectrometryXXXXXXMass spectroscopyXXXXXXMass spectroscopyXXXXXXIdentification of CarbohydratesXXXXXIdentification of sucrose, lactose and starchXXXXXSynthesis of 3.5-dimethyl pyrazoleXXXXXSynthesis of 5-nitrosalicylic acidXXXXXSynthesis of	$\begin{array}{c c c c c c c } \hline \mathbf{reactions} & \mathbf{N} & $	reactionsAAABACCC	reactionsII

	tetrahydrobenzo(b)thiophene-3- carboxylate								
12	Activity (Synthesis of certain drugs containing heterocycles)					X	X	X	X

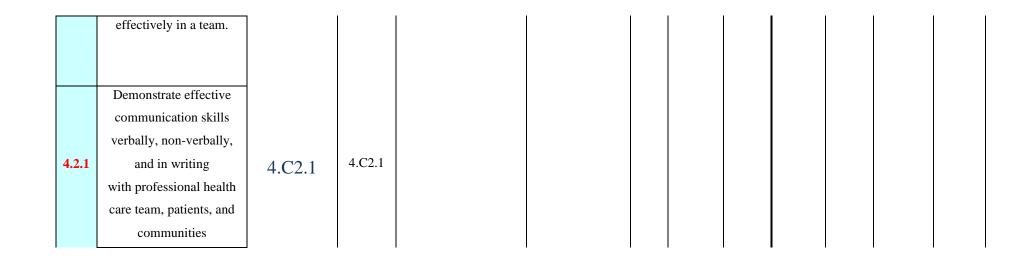
	Matrix II of Pharmaceutical organic chemistry-3													
Na	tional Academic	nal Academic Program		Course		Teaching and learning methods				Weighting of assessment				
Ref	erence Standards NARS	key elements	e key eleme nts	Course contents		Sources	Sources	lecture	practical session	self learning	written exam	practical exam	oral exam	Midterm exam
1.1.1	Demonstrate understanding of knowledge of pharmaceutical, biomedical, social, behavioral, administrative, and clinical sciences.	1.C1.1	1.C1.1 1.C1.2 1.C1.3	1- Stereochemistry *Definition, aim of study and classification *Structural isomerism *Rotational isomerism Geometrical isomerism Optical isomerism Activity (QUIZ) * D, L and Erythro, Threo Nomenclature *R and S, Enantiomers , Diastereomers * Activity (QUIZ) 2- Heterocyclic chemistry Nomenclature of heterocyclic & fused heterocyclic compounds Five member ring Furan, Pyrrole&Thiophenes ynthsis. Five membered ring Furan,	Student book Essential books Recommended books	x			X		X	X		

1			1		
	Pyrrole&Thio	hene			
	Reactions				
	*Activity (Qui	Z)			
	Five membere				
	with two nitro	gens			
	Pyrazole& Imi	dazole			
	Synthesis &				
	Properties				
	Indole synthes	s,			
	reaction and				
	serotonin (5H)))			
	Six-membered				
	Pyridine and it				
	derivatives, qu				
	and isoquinoli				
	Synthesis &				
	Reactions.				
	Six-membered	ring			
	with two nitro				
	Pyrimidine,				
	Pyrazine&Pyr	dazine			
	Synthesis &				
	Properties, pur	ine			
	nucleus				
	3- Carbohydra				
	Introduction a				
	Classification.				
	Monosacchari	les			
	i. Synthesis				
	ii.Cyclic struct	are of			
	Monosacchari	les			
	iii.Chemical				
	reactions of				
	Monosacchari	les			
	Activity (case				
	on nomenclatu				
	heterocycles)				
	Disaccharides				
	i.Nomenclatur				
	ii. Chemical				
	reactions of				
	reactions of				

				Disaccharides *.Polysaccharides i.Nomenclature ii.Chemical reactions and determination of the type of glycosidic linkage. *Activity (quiz)					
2.2.1	Isolate, design, identify, synthesize, purify, analyze, and standardize synthetic/ natural pharmaceutical materials.	2.C2.1.	2.C2.1 2.C2.2 2.C2.3 2.C2.4	Lab. Safety Identification of Carbohydrates Identification of glucose and fructose Identification of Carbohydrates Identification of sucrose, lactose and	Student book Essential books	X		x	
2.2.3	Recognize the principles of various tools and instruments, and select the proper techniques for synthesis and analysis of different materials and production of pharmaceuticals.	2.C2.4	2.C2.1 2.C2.4	starchSynthesis offructosazoneSynthesis of β-penta-acetyl glucose1Synthesis of 3,5-	Recommended books Practical note	x		x	
2.3.1	Handle, identify, and dispose biologicals, synthetic/natural materials, biotechnology-based and radio-labeled products, and other materials/products used	2.C3.1.	2.C3.1	quinoxalin-2-one Synthesis of 1,2,3- benzotriazole Synthesis of ethyl 2- amino-4,5,6,7- tetrahydrobenzo(b)th iophene-3- carboxylate Activity (Synthesis of certain drugs containing heterocycles)	Practical notes	X		x	

	in pharmaceutical field.		
2.3.2	Recognize and adopt ethical, legal, and safety guidelines for handling and disposal of biologicals, and pharmaceutical materials/products.	2.C3.2	2.C3.2
4.1.1	Demonstrate responsibility for team performance and peer evaluation of other team members, and express time management skills	4.C1.1	4.C1.1
4.1.2	Retrieve and critically analyze information, identify and solve problems, and work autonomously and	4.C1.5	4.C1.2

Practical notes	X		x	
Practical notebook	X		X	



COURSE SPECIFICATIONS

Pharmaceutical Analytical Chemistry III

Second Level– Semester 3

2020-2021

Course Specification of Pharmaceutical Analytical Chemistry III

University:	Zagazig	Facu	ltv: F	harmacy
A- Course spec	0 0	T ucu	1. j. 1	harmey
Program(s) on whi	ch the course is g	iven: Bachelor	of Pharmacy	(Pharm D)
Major or Minor ele	ement of program	s: Major		
Department offerin	ng the program:			
Department offerin	ng the course:	Analytic	al Chemistry	Į
Academic year / L	evel:	Second level	/ third semes	ster
Date of specification	on approval	2020		
B- Basic inform	nation:			
Title: Pharmaceuti	cal Analytical Ch	emistry III		PA 303
Credit Hours:				
Lectures : 1 hrs/we	ek			
Practical: 1 hrs/we	ek			
Tutorials:				
Total: 2 hrs/week				

C- Professional information:

1-Overall Aims of the Course:

On completion of the course, students will be able to:

- Explain the features of environmental analytical chemistry and various methods of sampling, samples preparation and analysis of different environmental samples, the physical and chemical composition of water, fat and oil.
- Apply quantitative methods for determination of different impurities in environment.

2- Key elements of Pharmaceutical Analytical Chemistry III :

DOMAIN 1- FUNDAMENTAL KNOWLEDGE

1-1- COMPETENCY:

Integrate knowledge from basic and applied pharmaceutical and clinical sciences to standardize materials, formulate and manufacture products, and deliver population and patient-centered care.

1.C1.1	Explain the importance of environmental analysis							
1.C1.2.	- Describe various methods of sampling, samples preparation							
	and standardization of water, oil and fat							
1.C1.3.	- Demonstrate the application of different techniques in							
	quantitative analysis of different pollutants, fat and oil.							

DOMAIN 2: PROFESSIONAL AND ETHICAL PRACTICE

2-2- COMPETENCY:

Standardize pharmaceutical materials, formulate and manufacture pharmaceutical products, and participate in systems for dispensing, storage, and distribution of medicines.

	2 (2 1	Perform chemical reactions for determination of some pollutants in air and soil.
2.C2.1.	pollutants in air and soil.	
ſ	2 (22 2	

2.C2.2 .	Perform laboratory tests for examination of water, fat and oil.
2 (2 2	Select the most suitable method for preparation and analysis of different environmental pollutants in their matrices.
2.02.3.	of different environmental pollutants in their matrices.

2-3- COMPETENCY:

Handle and dispose biologicals and synthetic/natural pharmaceutical materials /products effectively and safely with respect to relevant laws and legislations.

2.C3.1. Handle and dispose chemicals safely.

DOMAIN 4: PERSONAL PRACTICE

4-1- COMPETENCY

Express leadership, time management, critical thinking, problem solving, independent and team working, creativity and entrepreneurial skills.

- **4.C1.1.** Work as member of team.
- **4.C1.2.** Manage time and perform a task within time limit.
- **4.C1.3.** Develop problem solving skills.

D- Contents:

Week	Lecture	Practical session				
No.	(1 hrs/week)	(2 hr/week)				
1	- The importance of environmental	- Determination of O_3 and NO_2				
-	analytical chemistry.	(Demonstration)				
	- Different methods of sampling and	- Determination of Chloride in water				
2	sample preparation of environmental	(Mohr method)				
	pollutants samples.					
3	- Different types of chemical pollutants	- Limit test				
3	of air and soil and techniques for their determination.	- Emili test				
		T • • • • •				
4	- Physical examination of water	- Limit test				
5	- Chemical examination of water	- Determination of water alkalinity				
6	- Metals in water and interpretation of	-Determination of water hardness				
	analytical results	(complexometric method)				
7	Periodical Exam					
8	- Water pollution	-Determination of water hardness				
		(soap method)				
9	- Purification of water	- Colour test of oil				
10	- Physical properties, composition and	- Colour test of oil				
10	classification of oil and fat					
11	- Chemical properties of oil and fat	Activity				
12	- Rancidity	- Revision				
13	- Hydrogenation and analysis of butter	Practical exam				
15	fat					
14	- General discussion and revision					
15	Final Exam					

E- Teaching and Learning Methods:

- Lectures (data show, board)
- Practical sessions
- Problem solving
- Blended learning

F- Student Assessment Methods:

1- Written exam	to assess	1.C1.1,1.C1.2,1.C1.3,2.C2.3,4.C1.3
2- Practical exam	to assess	2.C2.1,2.C2.2, 2.C3.1
3- Oral exam	to assess	1.C1.1, 1.C1.2, 1.C1.3, 2.C2.3
4- Periodical exam	to assess	1.C1.1, 1.C1.2, 1.C1.3, 2.C2.3
5-Activities	to assess	4.C1.1, 4.C1.2

Assessment schedule:

Assessment (1):Final written exam	Week 15
Assessment (2): Practical exam	Week 13
Assessment (3): Oral exam	Week 15
Assessment (4):periodical exam	Week 7
Assessment (5): Activities	Week 11

Weighting of Assessment:

Assessment method	Marks	Percentage
Final written exam	50	50%
periodical exam	10	10%
Practical exam	25	25%
Oral exam	10	10%
Activity	5	5%
TOTAL	100	100%

G- Facilities Required for Teaching and Learning:

- Lectures (Computer, board)
- Practical sessions (Chemicals and Glassware)
- Activity
- Open discussion

H-List of References:

1- Course Notes: Student book "Pharmaceutical analytical chemistry III" approved by the analytical chemistry department book ,2020

2- Essential (textbooks):

i- J. Mendham, et al., Vogel's Textbook of Quantitative Chemical Analysis (6th edition);, Addison Wesley Publishing Co., 2000

ii- Daniel C. Harris, Quantitative Chemical Analysis (6th Edition);.(2002).

iii. Roger N. Reeve. Introduction to Environmental Analysis(1st Edition);Wiley publishing., (2002).

3- Recommended books:

- D. C. Harris, Quantitative Analytical Chemistry (9th edition), W. H.
 Freeman and Co. (2015)
- D. Chowrasia, N. Sharma, Analytical Chemistry. A Qualitative & Quantitative Approach (General Techniques) Knoc education (2015).
- iii. O. Hutzinger, et al. The Handbook of Environmental Chemistry. Volume 19.

4- Periodicals, Web Sites, etc

https://www.ekb.eg/ http://chemwiki.ucdavis.edu/ http://en.wikipedia.org/ www.Pubmed.Com and www.sciencedirect.com

Course Coordinator: Prof. Dr.

Head of Department: Prof. Dr. Hisham Ezzat

تم مناقشة واعتماد توصيف المقرر من مجلس القسم بتاريخ Date: 2020

	Matrix I of Analytical Chemistry 3 course												
	Course Contents		1-1- IPETEI	NCY	CON	2-2- MPETE	NCY	2-3- COMPETENCY		4-1- COMPETENCY			
			1.C1	1.C1	2.C2.	2.C2.	2.C2.	2.C3.1	4.C1.	4.C1.2	4.C1.		
Lectures		1	.2	.3	1	2	3	2.03.1	1	4.01.2	3		
1	- The importance of environmental analytical chemistry	Х											
2	- Different methods of sampling and sample preparation of environmental pollutants samples.		X										
3	- Different types of chemical pollutants of			х			х						
4	- Physical examination of water		Х										
5	- Chemical examination of water		х										
6	6 - Metals in water and interpretation of analytical results		х	х									
7 Midterm exam													
8	- Water pollution		Х	Х									
9	- Purification of water		Х										

10	- Physical properties, composition and classification of oil and fat		х									
11	- Chemical properties of oil and fat		Х									
12	- Rancidity		Х	х								
13	- Hydrogenation and analysis of butter fat		х	х								
14	- General discussion and revision	Х	Х	х								
	Practical sessions		1.1.2	1.1.3	2.2.1	2.2.2	2.2.3	2.3.1	4.1.1	4.1.2	4.1.3	
1	- Determination of O ₃ and NO ₂ (Demonstration)				Х			Х	X		х	
2	-Determination of chloride in water (Mohr method)					х		Х	х		x	
3	-Limit test					х		Х	х		х	
4	- Limit test					Х		X	Х		х	
5	- Determination of water alkalinity					х		X	Х		х	
6	-Determination of water hardness (complexometric method)					x		Х	Х		x	
7	midterm Exam											
8	-Determination of water hardness (soap method)					х		Х	х		х	
9	- Colour test of oil					Х		Х	Х		х	
10	- Colour test of oil					Х		Х	Х		х	
11	activity								Х	Х	Х	

	Matrix II of Analytical Chemistry 3 course											
National Academic Reference		Program	Course key	Course contents	Sources	Teach	Teaching and learning methods Method			d of asses	of assessment	
Stan	idards (NARS)	key elements	elements			Lecture	Practical session	Self learning	Written exam	Practical exam	Oral exam	
1.1.1	Demonstrate understanding of knowledge of pharmaceutical, biomedical, social, behavioral,	al, 1.C1.1 nd	1.C1.1	 The importance of environmental analytical chemistry Different types of chemical pollutants of air and soil. Physical and chemical 	Student book Essential books Recommended books Internet	x		X	X		x	
	administrative, and clinical sciences.		1.C1.2.	 examination of water Water pollution Physical properties and classification of oil and fat Chemical properties of oil , fat Rancidity, hydrogenation and analysis of butter fat 								

1.1.3	Integrate knowledge from fundamental sciences to handle, identify, extract, design, prepare, analyze, and assure quality of synthetic/ natural pharmaceutical materials/products.	1.C1.9	1.C1.3	 Different methods of sampling and sample preparation of environmental pollutants samples. Physical and chemical examination of water Water pollution Physical properties and classification of oil and fat Chemical properties of oil, fat Rancidity, hydrogenation and analysis of butter fat 	Student book Essential books Recommended books Internet	х		х	X		x
	Isolate, design,		2.C2.1	- Determination of O ₃ and NO ₂							
2.2.1	identify, synthesize, purify, analyze, and standardize synthetic/ natural pharmaceutical materials.	2.C2.1.	2.C2.2	 -Determination of chloride in water (Mohr method) -Limit test - Determination of water alkalinity and hardness - Determination of saponification value Colour test of oil 	Practical sessions		х			Х	
2-2-3	Recognize the principles of various tools and instruments, and	2.C2.5.	2.C2.3	- Different techniques for determination of chemical pollutants of air and soil and	Student book Essential books Recommended books Internet	x		x	x	x	x

	select the proper techniques for synthesis and								
	analysis of different materials and production of pharmaceuticals.								
2-3-1	Handle, identify, and dispose biologicals, synthetic/natural materials, biotechnology- based and radio- labeled products, and other materials/products used in pharmaceutical field.	2.C3.1.	2.C3.1	- Practical sessions	Practical notes	X		X	
4-1-1	Demonstrate responsibility for	4.C1.1.	4.C1.1	- Practical sessions - Activity	Practical notes	X		X	

	members, and express time management skills	4.C1.3	4.C1.2						
4-1-2	Retrieve and critically analyze information, identify and solve problems, and work autonomously and effectively in a team.	4.C1.5.	4.C1.3	- Practical sessions Activity	Practical notes	X		X	

Course Coordinator: Prof. Dr. Gamal Ragab

Head of Department: Prof. Dr. Hisham Ezzat

تم مناقشة و اعتماد توصيف المقرر من مجلس القسم بتاريخ 2020

COURSE SPECIFICATIONS

Pharmacognosy II

Second Level– Semester 3

2020-2021

Course Specification of Pharmacognosy II

University:	Zagazig	Faculty:	Pharmacy
A- Course sp	ecifications:		
Program(s) on v	which the course is giv	en: Bachelor of Pharr	nacy (Pharm D).
Major or Minor	element of programs:	Major	
Department off	ering the program:		
Department off	ering the course:	Pharmacognosy	
Academic year	/Level:	second level /third s	emester
Date of specific	cation approval:	25 September 2020	0
B- Basic info	rmation:		
Title: Pharmaco	ognosy II	Code: PG303	
Credit Hours: S	ingle, credit hour syste	em	
Lectures: 2 hrs/	week		
Practical:1 hrs/v	week		
Tutorials: -			
Total: 3 hrs/we	ek		

C- Professional information:

1. Overall Aims of the Course:

On completion of the course, students will be able to illustrate microscopical and macroscopical characters and uses of medicinal flowers, barks, woods and seeds as well as identification of their different active constituents and adulteration.

2- Key elements of Pharmacognosy II

DOMAIN	N 1- FUNDAMENTAL KNOWLEDGE
clinical sc	IPETENCY: Integrate knowledge from basic and applied pharmaceutical and ciences to standardize materials, formulate and manufacture products, and deliver n and patient-centered care.
1.C1.1.	Describe morphological and histological characters of medicinal herbs, fruits, subterranean organs.
1.C1.2.	Identify different active constituents of medicinal uses of, medicinal herbs, fruits, subterranean organs.
1.C1.3.	Outline adulteration of different medicinal medicinal herbs, fruits, subterranean organs and unorganized plant and animal drugs.
DOMAIN	N 2: PROFESSIONAL AND ETHICAL PRACTICE
	MPETENCY: Standardize pharmaceutical materials, formulate and manufacture utical products, and participate in systems for dispensing, storage, and distribution nes.
2.C2.1.	Handle microscope efficiently and design protocols to examine medicinal plants.
2.C2.2.	Use equipments effectively.
2.C2.3.	Examine plant organs in entire and powdered form.
2.C2.4	Determine the active constituents of the studied drugs qualitatively.
DOMAIN	N 4: PERSONAL PRACTICE
Express le team work	IPETENCY eadership, time management, critical thinking, problem solving, independent and king, creativity and entrepreneurial skills.
4.C1.1.	Work effectively as part of a team.
4.C1.2.	Manage time and plan for work.
	IPETENCY: Effectively communicate verbally, non-verbally and in writing with Is and communities.
4.C2.1	Write and present reports

D- Contents:

Week	Lecture (2 hrs/week)	Practical session (1hr/week)
No 1	General introduction for what will be taught all	Laboratory Safety Measures
•	over the term	Dealing With Microscope.
	Introduction for the fruits and giving the students	Morphology of some important fruits
	the possible references, web sites, text books.	1 07 1
	Description of the common characters of	
	umbelliferous.	
2	Description of morphological and histological	Identification of Fennel in entire and
	characters, constituents, uses, chemical tests and	powdered form.
	detection of adulteration in entire and powdered	
	form of Fennel, Ammi visnaga and Ammi majus.	
3	Description of morphological and histological	Identification of Anise and Ammi visnaga,
	characters, constituents, uses, chemical tests and	Ammi majus and coriander in entire and
	detection of adulteration in entire and powdered	powdered form.
	form of Coriander, Anise, , Capsicum, Colocynth,	Identification of Capsicum and colocynth in
	Lemon, bitter orange peel and Senna pods.	entire and powdered form.
4	Evening primrose, Colchicum and mustard macro-	Mustard and nux vomica: macro-, and
	and, micro-morphology of the entire and	Micro-morphology, powders and chemical identification
	powdered drugs, chemical identification.	
	Unofficial fruits.	
_		pharmaceutical fruits)
5	Introduction to herbs.	Morphology of some important herbs.
	Study morphological and histological characters, constituents, uses, chemical tests and detection of	Identification of Mentha and Thyme.
	adulteration in entire and powdered form of	
	Lobelia.	
6	Study Morphological And Histological	Identification of Vinica
	Characters, Constituents, Uses, Chemical Tests	Activity (report on medical uses of common
	And Detection Of Adulteration In Entire And	herbs)
	Powdered Form Of Piperment, Thyme, Alfalfa	
	and Echinaceae. Unofficial herbs	
7	Midterm	
8	Introduction to subterranean organs.	Morphological demonstration for some
	Activity.	important roots and rhizomes. Identification
	Study morphological and histological characters,	of Liquorice
	constituents, uses, chemical tests and detection of	
	adulteration in entire and powdered form of,	
0	Liquorice	Maana mambalaanu misna mambalaan
9	Study morphological and histological characters, constituents, uses, chemical tests and detection of	Macro-morphology; micro-morphology
		powder and chemical identification of Ginger and Rhubarb.
	adulteration in entire and powdered form of senega	Uniger and Khubard.

	and Ipeca	
10	Study morphological and histological characters,	Identification of un-organized drugs(
	constituents, uses, chemical tests and detection of	morphology and chemical tests
	adulteration in entire and powdered form of	
	Rhubarb, Ginger.	
11	Study morphological and histological characters,	Revision
	constituents, uses, chemical tests and detection of	
	adulteration in entire and powdered form of	
	Hydrastis, Valeriana., Unofficial rhizomes	
12	Introduction to Unorganized drugs	Practical exam
	Study the preparation, collection, active	
	constituents, uses and chemical tests of resins,	
	oleo-resins	
13	Study preparation, collection, active constituents,	Practical exam
	uses and chemical tests of balsams, latex, juice,	
	extracts and gum.	
14	Revision	
15		
	Written and oral exam	

E- Teaching and Learning Methods:

- Lectures (interactive lecture, data show and board)
- Practical sessions
- Self-learning (Group discussion, Group assignment)
- Blended learning

F- Student Assessment Methods

- 1- Written exam to assess 1.C1.1, 1.C1.2, 1.C1.3
- 2- Practical exam to assess 2.C1.1, 2.C1.2, 2.C1.3 and 2.C1.4
- 3- Oral exam to assess 1.C1.1, 1.C1.2, 1.C1.3

4- Activities (group discussion, presentation, net search and pamphlets' of natural

drugs) to assess 4.C2.1, 4.C1.1 and 4.C1.2

Assessment schedule:

Assessment (1): midterm	Week 7
Assessment (2):Final Written exam	Week 15
Assessment (3): Activity(group	
discussion, presentation, net search	Week 6

and pamphlets' of natural drugs)	
Assessment (4): Activity(group	Week 11
discussion, presentation, net search	
and pamphlets' of natural drugs)	
Assessment (4): Practical exam	Week 13
Assessment (5): Oral exam	Week 15

Weighting of Assessment:

Assessment method	Marks	Percentage
Written exams	50	50%
Mid term	10	10%
Activity	5	5%
Practical exam	25	25%
Oral exam	10	10%
TOTAL	100	100%

G- Facilities Required for Teaching and Learning:

• For lectures: Black (white) boards, data show, air conditioned classroom

• For practical: Well-equipped labs

- Chemicals e.g. Hcl, KoH, glycerin and phloroglucinol, glassware, slides, covers,

digital balances, flame and water baths

H- List of References:

1- Student's book approved by Pharmacognosy Department on 2019.

2- Essential Books:

- pharmacognosy, fundmentals, applications and strategies by simone Badal and Rupika Delgoda (2017)

-Trease and Evans, Pharmacognosy, 15thed., Saunders company, Nottingham,U.K. Willium Charles Evans (2003).

3- Recommended Books

Leung A.Y. and Faster" Encyclopedia of Common Natural Ingredients Used in Food, Drugs and Cosmetics".

- Janice, Glimn-Lacy and Peter B. Kaufman, Botany Illustrated, Introduction to plants, major groups, flowering plants families, 2nd ed. Springer **2006**.
- Martindale, "The extrapharmacopeia". 31st Edn., by James, E.F Reynolds. And Kathleen Parfitt, Royal Pharmaceutical Society, London (2007).

4- Periodicals, web sites, etc

Amer. J. Nat. Prod., Phytochemistry, Planta Medica, Fitoterapia.

- A. Fahan, Plant Anatomy, Pergamon Press. 2002.
- http://www.scribd.com/doc/75980088/Atlas-of-Medicinal-Plants-II
- http://pharmacystudent-prep.blogspot.com
- http://www.pharma-board.com/board/fopgal/index.php

https://www.google.com/search?safe=active & sxsrf=ACYBGNT1wfCQl6DGxZ5ouZYl1QZZfJSrYindersearch?safe=active & sxsrf=ACYBGNT1wfCQl6DGxZ5ouZYl1QZsfJSrYindersearch?safe=active & sxsrf=active & sxsrf=ACYBGNT1wfCQl6DGxZ5ouZYl1QZzfJSrYindersearch?safe=active & sxsrf=ACYBGNT1wfCQl6DGxZ5ouZYl1QZsfJSrYindersearch?safe=active & sxsrf=ACYBGNT1wfCQl6DGxZ5ouZYl1QZsfJSrYindersearch?safe=active & sxsrf=ACYBGNT1wfCQl6DGxZ5ouZYl1QZsfJSrYindersearch?safe=active & sxsrf=ACYBGNT1wfCQl6DGxZ5ouZYl1QZsfJSrYindersearch?safe=active & sxsrf=ACYBGNT1wfCQl6DGxZ5ouZYl1QZsfJSrYindersearch?safe=active & sxsrf=ACYBGNT1wfCQl6DGxZ5ouZYl1QZsf

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Course Coordinators: Prof. Dr. Afaf abd El-Ghany

Head of department: Prof. Dr. Amal Al-Gendy

تم مناقشة و إعتماد توصيف المقرر من مجلس القسم بتاريخ /2020/9 Date: 2020/9

			Ma	trix I o	f Phai	maco	gnosy	-II cou	rse				
	C			Key	y elem	ents of	Pharm						
	Course Contents	1-1- COMPETENCY			CO	2-2- COMPETENCY			4-1- COMPETENCY			4-2- COMPETENCY	
	Lectures	1.C1.1	1.C1.2	1.C1.3	2.C2.1	2.C2.2	2.C2.3	2.C2.4	4.C1.1	4.C1.2	4.C	2.1	
1	General introduction for what will be taught all over the term Introduction for the fruits and giving the students the possible references, web sites, text books. Description of the common characters of umbelliferous.	*											
2	Description of the common characters of umbelliferous. Description of morphological and histological characters, constituents, uses, chemical tests and detection of adulteration in entire and powdered form of Ammi visnaga and Ammi majus.	*	*	*									
3	Description of morphological and	*	*	*									

	histological							
	characters,							
	constituents,							
	uses, chemical							
	tests and							
	detection of							
	adulteration in							
	entire and							
	powdered							
	form of							
	Coriander,							
	Anise, Fennel,							
	Capsicum,							
	Colocynth,							
	Lemon, bitter							
	orange peel							
	and Senna							
	pods.							
	Evening							
	primrose,							
	Colchicum							
	and mustard							
	macro-and,							
	micro-							
	morphology of	*	ala	ala				
4	the entire and	*	*	*				
	powdered							
	drugs,							
	chemical							
	identification.							
	Unofficial							
	fruits.							
	Introduction							
	to herbs.							
	Study							
	morphological							
	and							
	histological							
5	characters,	*						
3								
	constituents,							
	uses, chemical							
	tests and							
	detection of							
	adulteration in							
	entire and							

	h and and							
	powdered							
	form of							
	Lobelia.							
6	Study Morphological And Histological Characters, Constituents, Uses, Chemical Tests And Detection Of Adulteration In Entire And Powdered Form Of Piperment, Thyme, Alfalfa and Echinaceae. Unofficial herbs	*	*	*				
7	Introduction to subterranean organs. Study morphological and histological characters, constituents, uses, chemical tests and detection of adulteration in entire and powdered form of, Liquorice	*	*	*				
8	Study morphological and histological	*	*	*				

	dmaaa	<u> </u>								
	drugs									
	Study the									
	preparation,									
	collection,									
	active									
	constituents,									
	uses and									
	chemical tests									
	of resins,									
	oleo-resins									
	Study									
	preparation,									
	collection,									
	active									
	constituents,									
12	uses and	*	*	*						
	chemical tests									
	of balsams,									
	latex, juice,									
	extracts and									
	gum.									
	Practical sess	ion					1			
	Laboratory									
	Safety									
	Measures									
	Dealing With									
1	Microscope.				*					
	Morphology									
	of some									
	of some important									
	important									
	important fruits									
	important fruits Identification									
2	important fruits Identification of Fennel in					*				
2	important fruits Identification of Fennel in entire and					*				
2	important fruits Identification of Fennel in entire and powdered					*				
2	important fruits Identification of Fennel in entire and powdered form.					*				
2	important fruits Identification of Fennel in entire and powdered form. Identification					*				
2	important fruits Identification of Fennel in entire and powdered form. Identification of Anise and					*				
2	important fruits Identification of Fennel in entire and powdered form. Identification of Anise and Ammi					*				
	important fruits Identification of Fennel in entire and powdered form. Identification of Anise and Ammi visnaga,					*				
2	important fruits Identification of Fennel in entire and powdered form. Identification of Anise and Ammi visnaga, Ammi majus									
	important fruits Identification of Fennel in entire and powdered form. Identification of Anise and Ammi visnaga, Ammi majus and coriander									
	important fruits Identification of Fennel in entire and powdered form. Identification of Anise and Ammi visnaga, Ammi majus and coriander in entire and									
	important fruits Identification of Fennel in entire and powdered form. Identification of Anise and Ammi visnaga, Ammi majus and coriander									

			-	1	1	1			1		
	Identification										
	of Capsicum										
	and colocynth										
	in entire and										
	powdered										
	form.										
	Mustard and										
	nux vomica:										
	macro-, and										
	Micro-										
	morphology,										
4	powders and				*			*			
	chemical										
	identification										
	Activity (report										
	on different										
	pharmaceutical fruits)										
	Morphology										
	of some										
	important										
5	herbs.				*						
	Identification										
	of Mentha and										
	Thyme.										
	Identification										
	of Vinica										
	Activity				*			*			
6	(report on										
	medical uses										
	of common										
	herbs)										
	Morphological										
	demonstration										
	for some										
	important										
7	roots and								*	*	*
	rhizomes.										
	Identification										
	of Liquorice										
	Macro-										
	morphology;					*	*	*			
8	micro-										
	morphology										
	1 07			1	1	1	I			1	

Teaching and learning methods Program Weighting of a demic Course **Course contents** Sources lecture practical Activities written practical key key e (

Matrix II of Pharmacognosy-II course

	powder and chemical identification of Ginger and Rhubarb.								
9	Identification of un- organized drugs(morphology and chemical tests				*	*			
10	Revision.				*	*	*	*	*

	elements	elements				session	reports and presentations) and Field visit	exam	exam & activity
ARS									
		1.C1.1	- Overview on the course in pharmacognosy II	Student book Essential books	*			*	
nstrate anding of edge of ceutical, cal, social, vioral, ative, and sciences.	1.C1.2	1.C1.2	-Morphological and histological studies for plant fruits.herbs,subterenean organs in entire and powdered forms, active constituents, uses and chemical test.	Student book Essential books	*			*	
		1.C1.3	adulteration of different medicinal fruits,herbs,subterranean organs	Student book Essential books	*			*	
grate lage from mental to handle, extract, prepare, nd assure ty of	1.C1.9	1.C1.2	- Morphological and histological studies for plant fruits,herbs,subterranean organs in entire and powdered forms, active constituents, uses and chemical test.	Student book Essential books	*	*	*	*	*
c/ natural ceutical /products.		1.C1.3	adulteration of different medicinal fruits,herbs and subterranean organs	Self learning					
		2C.2.1	Uses of microscopes	Practical notes		*			*
design, tify,		2.C2.2	Microscopical examination of different flowers	Practical notes		*			*
e, purify, e, and ardize :/ natural	2.C2.1	2.C2.3	Morphological and histological studies for plant fruits and herbs in entire and powdered forms	Practical notes		*			*
ceutical erials.		2.C2.4	Morphological and histological studies for subterranean organs in entire and powdered forms	Practical notes		*			*
nstrate ibility for		4.C1.1					*		*
formance peer n of other embers, ress time nent skills.	4.C1.1 4.C1.3	4.C12	Activity 1 (researches and reports on fruits and herbs with pharmaceutical preparations Activity 2 (researches and	Self learning			*		*
emporary gies and ia to nstrate ctive cion skills.	4.C2.2	4.C2.1	presentation on pharmaceutical preparations containing subterranean organs)				*		*

Course Coordinators: Prof. Dr. Afaf abd El-Ghany

Head of department: Prof. Dr. Amal Al-Gendy

تم مناقشة و إعتماد توصيف المقرر من مجلس القسم بتاريخ /2020/9 Date: 2020/9

COURSE SPECIFICATIONS

Biophysics

Second Level– Semester 3 2020-2021

Course Specification of Biophysics for (2020/2021)

University: ZagazigFaculty: PharmacyA- Course specifications:

Program(s) on which the course is given: Bachelor of pharmacy program(Pharm D)

Major or Minor element of programs:MajorDepartment offering the program:------Department offering the course:Biochemistry, Faculty of PharmacyAcademic year Level:Second level /Semester 3Date of specification approval:

Code: MD 303

B- Basic information:

Title: Biophysics

Credit hours:

- Lectures: 1 hr/week
- Practical: 1 hr/week
- Tutorials: ---
- Total: 2 hrs/week

C- Professional information:

<u>1-Overall Aims of the Course:</u>

On completion of the course, students will be able to:

Explain biological phenomena using the principles and techniques of physics and modify these techniques to help diagnosis and treat various disorders

2- Key elements of Biophysics:

DOMAD										
DOMAI	N 1- FUNDAMENTAL KNOWLEDGE									
	1-1- COMPETENCY: Integrate knowledge from basic and applied pharmaceutical									
and clinical sciences to standardize materials, formulate and manufacture products,										
	and deliver population and patient-centered care.									
1.C1.1.	Explain the principles of biophysics									
Outline the structure of cell membrane and mechanisms o										
1.C1.2	transport across the cell membrane.									
	transport across the cert memorane.									
1.C1.3 Identify mechanisms of signal transduction and mechanism of										
action of different types of receptors.										
1.01.4	Illustrate the bases of biophysical techniques as ECG, Laser									
1.C1.4	and Radiation and their different applications									
	and Radiation and then different applications									
1.01.5	Use the proper abbreviations and symbols related to									
1.C1.5	biophysics									
	biophysics									
DOMAI	N 3: PHARMACEUTICAL CARE									
3-1- CON	IPETENCY: Apply the principles of body functions to participate in									
improving	g health care services using evidence-based data.									
3.C1.1.	Link between membrane defect and related disorders.									
3.C1.2	Interpret biophysical measurements including blood pressure									
	ECG									
DOMAI	N 4: PERSONAL PRACTICE									
4-1- CON	IPETENCY Express leadership, time management, critical									
	, problem solving, independent and team working, creativity									
-	epreneurial skills.									
	-									
	Recognise the value and structure of the pharmacy team and of									
4.C1.1.	a multiprofessional team									

Week	Lecture	Practical session
No.	(1hr/week)	(1hrs/week)
1	-Structure of the plasma membrane	-Lab safety procedures guidelines
2	-Transport across the plasma membrane 1	-Atomic Physics
3	-Transport across the plasma membrane 2	-Biochemical Bonds
4	-Channels and carriers	-The Plasma membrane
5	-Signal Transduction and receptors	-Transport across the plasma membrane
6	- Self learning activity (diagnostic uses of x-ray)	- Practical exam 1
7	-periodical exam	Activity(application of radioactive isotopes in medical field presented as report and ppt)
8	-Biophysical basis of ECG and blood pressure measurement (electrochemical gradient and membrane action potential – action potential in the heart)	-Ion channel
9	- Action potential in the heart (cont.) (ECG technique and interpretation – blood pressure measurement)	-Receptors
10	-Atom and radiation (the quantum model of the atom – electromagnetic spectrum – radioactive decay – types of radiation)	-Water homeostasis
11	-Quantification of radiation (sources of radiation – radiation risks – application of radiology)	-Blood pressure and viscosity
12	Laser technology (laser beam properties and generation)	-Action potential
13	- Laser types , hazards and applications	-Heart electricity and ECG
14	- Revision and open discussion	-Practical exam 2
15	-Written exam	

- 2
- Interactive lectures
- Practical session
- Research assignment
- Self learning
- Blended learning

F- Student Assessment Methods:

1-Written exams to assess: 1.C1.1, 1.C1.2, 1.C1.3, 1.C1.4, 1.C1.5

- 2- Practical exams to assess: 3.C1.1, 3.C1.2
- 3- Activities to assess: 4.C1.1

4- Oral exam to assess: : 1.C1.1, 1.C1.2, 1.C1.3, 1.C1.4, 1.C1.5

Assessment schedule:

Assessment (1): Written exam	Week 15
Assessment (2): Practical exams	Week 6,14
Assessment (3): Periodical exam	Week 7
Assessment (4): Oral exam	Week 15
Assessment (5): Activity	Week 6, 7

Weighting of Assessment

Assessment method	Marks	Percentage
Written exam	50	50%
Practical exam	25	25%
Periodical exam	10	10%
activity	5	5%
Oral exam	10	10%
TOTAL	100	100%

<u>G-Facilities Required for Teaching and Learning:</u>

• Data show , software , videos and screens.

H- List of references:

- **1- Course Notes**
- Student book of Biophysics part1 approved by biochemistry department2020
- Student book of Biophysics part2 approved by biochemistry department2020

- Practical note of Biophysics approved by biochemistry department2020
- 2- Essential books
 - The biophysics of cell membranes-Epand , Richard M, Ruysschaert , Jean Marie (2017)
 - Introduction to experimental Biophysics, 2ndedition-Jayl Nadeau (2018)
 - Membrane biophysics : New insights and methods -HongdaWang , Guohuili (2018)

3- Recommended books

- Fundamentals of Biophysics Andrey B. Rubin, Wiley-Scrivener (2014)
- Biophysics and neurophysiology of the six sense NimaRezaei , AmeneSaghazadeh.(2019)
- Course Coordinators: Dr. Nahla Younis
- Head of Department: Dr. Nahla Younis
- تم مناقشة و إعتماد توصيف المقرر من مجلس القسم بتاريخ

		Ma	trix I o	f Bioph	ysics							
	Course Contents	Key elements of biophysics										
	Course Contents	1-1- COMPETENCY						3- PETENCY	4-1- COMPETENCY			
	Lectures		1.C1.2	1.C1.3	1.C1.4	1.C1.5	3.C1.1		4.C1.1			
1		1.C1.1	1.01.2	1.01.5	1.01.4		5.C1.1	5.01.2	4.01.1			
1	-Structure of the plasma membrane	X	X			X						
2	-Transport across the plasma membrane 1	X	X			X						
3	-Transport across the plasma membrane 2	X	X			X						
4	Channels and carriers	Х	X			X						
5	-Signal Transduction and receptors	Х		Х		Χ						
6	-Biophysical basis of ECG and blood pressure measurement (electrochemical gradient and membrane action potential – action potential in the heart)	Х			X	X						
7	 Action potential in the heart (cont.) (ECG technique and interpretation – blood pressure measurement) 	X			X	X						
8	-Atom and radiation (the quantum model of the atom – electromagnetic spectrum – radioactive decay – types of radiation)	X			X	X						
9	-Quantification of radiation (sources of radiation – radiation risks – application of radiology)	X			X	X						
10	Laser technology (laser beam properties and generation)	X			X	Х						
11	- Laser types, hazards and applications	X			X	X						
	Practical sessions											
1	-Lab safety procedures guidelines						X					
2	-Atomic Physics						X					

3	-Biochemical Bonds			Х		
4	-The Plasma membrane			Х		
5	-Transport across the plasma membrane			Х		
6	-Ion channel			X		
7	Receptors				X	
8	-Water homeostasis				X	
9	-Blood pressure and viscosity				X	
10	-Action potential				X	
11	- Self learning activity (diagnostic uses of x- ray)					X
12	Heart electricity and ECG				Х	X
13	Activity(application of radioactive isotopes in medical field presented as report and ppt)					X

	Matrix II of Biophysics											
N	ational Academic	Program	Cours	Course			aching learnin method	g	Weighting of assessment			
Re	ference Standards NARS	key elements	e key eleme nts	Course contents Sources		lecture	practical session	self learning	written exam	practical exam	oral exam	Midterm exam
1.1.1	Demonstrate understanding of knowledge of pharmaceutical, biomedical, social, behavioral, administrative, and clinical sciences.	1.C1.1	1.C1.1 1.C1.2 1.C1.3 1.C1.4	-Structure of the plasma membrane -Transport across the plasma membrane 1 -Transport across the plasma membrane 2 Channels and carriers -Signal Transduction and receptors -Biophysical basis of ECG and blood pressure	Student book Essential books Recommended books	x			X		X	X
1.1.2	Utilize the proper pharmaceutical and medical terms, abbreviations and symbols in pharmacy practice	1.C1.8	1.C1.5	measurement (electrochemical gradient and membrane action potential – action potential in the heart) - Action potential in the heart (cont.) (ECG technique and interpretation – blood pressure measurement) -Atom and radiation (the quantum model of the atom – electromagnetic spectrum – radioactive decay –	Student book Essential books Recommended books	x			X		X	

				types of radiation) -Quantification of radiation (sources of radiation – radiation risks – application of radiology) Laser technology (laser beam properties and generation) - Laser types , hazards and applications					
3.1. 3	Monitor and control microbial growth and carry out laboratory tests for identification of infections/ diseases.	3.C1.1	3.C1.1 3.C1.2	-Atomic Physics -Biochemical Bonds -The Plasma membrane -Transport across the plasma membrane -Ion channel -Receptors -Water homeostasis -Blood pressure and viscosity -Action potential - Self learning activity (diagnostic uses of x-ray) Heart electricity and ECG Activity(application of radioactive isotopes in medical field presented as report and ppt)	Student book Essential books Recommended books Practical note	X		X	
4.1.1	Demonstrate responsibility for team performance and peer evaluation of other team	4.C1.1	4.C1.1	 Self learning activity (diagnostic uses of x- ray) Activity(application of radioactive isotopes in medical 	internet		x		

	members, and express	field presented as				
	time management skills	report and ppt)				

COURSE SPECIFICATIONS

Physiology and Pathophysiology

Second Level– Semester 3 2020-2021

Course specification of Physiology & Pathophysiology

Unive	ersity:	Zagazig	Faculty:	Pharmacy
A- C	ourse sj	pecifications:		
Progr	am (s) on	which the course is gi	ven: Bachelor of Pha	rmacy (pharmD)
Majo	r or Mino	r element of programs	: Major	
Depar	rtment of	fering the program:		
Depar	rtment of	fering the course:	Pharmacology and to	oxicology Department
Acade	emic year	Level:	Second level/ semes	ster 3
Date	of specifi	cation approval:	July 2020	
B-B	asic info	ormation:		
Title:	Physiolo	ogy & Pathophysiology	Co	ode: MD 304
Credit	t Hours: 3	5		
Lectu	res: 2hrs/	week		
Practi	ical: 1hrs/	week		
Tutor	ials: -			
Total	: 3 hrs/we	eek		

C- Professional information:

1-Overall aim of the course

On completion of the course, the student will be able to explain the basic principles of physiology of major organ systems and the pathophysiology of common diseases.

2- Key elements of Physiology & Pathophysiology

1-1- CO	1-1- COMPETENCY		
	Integrate knowledge from basic and applied pharmaceutical and clinical sciences to		
standardize materials, formulate and manufacture products, and deliver population and			
patient-centered care.			
1.C1.1	Outline the basic physiology of major organ systems.		
1.C1.2	Illustrate the pathophysiology of common diseases.		
1.C1.3	Outline drug classes for the treatment of certain diseases		
3-1- COMPETENCY Apply the principles of body functions to participate in improving health care services			
using evi	dence-based data.		
3.C1.1	Handle data effectively to recognize physiology/pathophysiology		
	of organ systems		
3.C1.2	Assess health profiles of patient cases		
3.C1.3	3 Suggest suitable drug classes for the treatment of certain diseases		
3.C1.4	1.4 Integrate information from different sources to solve clinical cases.		
4-1- COMPETENCY			
Express leadership, time management, critical thinking, problem solving, independent			
and team working, creativity and entrepreneurial skills.			
4.C1.1	Work effectively as a member of teamwork		

D- Contents:

Week	Lecture contents (2 hrs/week)	Practical session	
No.		(1hr/week)	
1	Cell physiology Neuromuscular and synaptic transmission	Lab safety rules	
2	PNS Physiology Nervous system physiology	Basic principles of case studies	
3	Central nervous system physiology	Pathophysiology of Psychosis Pathophysiology of Parkinsonism	
4	Central nervous system physiology	Pathophysiology of Alzheimer disease Pathophysiology of Epilepsy	
5	Endocrine physiology	Pathophysiology of Diabetes , other hormonal disorders	

6	Renal physiology	Pathophysiology of renal failure	
7	Midterm exam		
8	Respiratory physiology	Pathophysiology of Asthma Pathophysiology of Cough	
9	GIT physiology	Pathophysiology of Hepatitis, Ulcer, Diarrhea, Constipation	
10	Cardiovascular system physiology	Pathophysiology of Hypertension	
11	Cardiovascular system physiology	Pathophysiology of Heart failure Pathophysiology of Angina	
12	The lymphatic system and immunity	Pathophysiology of Inflammation	
13	The lymphatic system and immunity	Pathophysiology Infectious disease	
14	Revision and open discussion	Practical exam	
15	Final written exam		

E- Teaching and Learning Methods:

- Lectures
- Practical sessions
- Think/pair/share
- Case study
- Blended learning

F- Student Assessment methods:

- 1- Written exams to assess: 1.C1.1, 1.C1.2, 1.C1.3
- 2- Case study in team work to assess : 4.C1.1, 3.C1.4
- 3- Practical exams to assess: 3.C1.1, 3.C1.2, 3.C1.3, 3.C1.4
- 4- Oral exam to assess: 1.C1.1, 1.C1.2, 1.C1.3

Assessment schedule

Assessment (1):Midterm exam	Week 7	
Assessment (2):Activity (case studies in	Week 9,10	
team work)		
Assessment (3): Practical exam	Week 14	

Assessment (4): Final written exam	Week 15
Assessment (5): Oral exams	Week 15

Weighting of Assessment

Assessment method	Marks	Percentage
Midterm exam	10	10%
Activity (case studies in team	5	5%
work)		
Practical practice & exam	25	25%
Final written exam	50	50%
Oral exam	10	10%
TOTAL	100	100%

G- Facilities required for teaching and learning:

• For lectures : Black (white) boards, data show, air conditioned classroom

• For practical: Well-equipped labs and chemicals.

H- List of References:

1- Course Notes: Student book approved by Toxicology and Pharmacology

department (2020)

- Practical notes approved by Toxicology and Pharmacology department (2020)

2- Essential Books:

i- Essentials of Pathophysiology: Concepts of Altered Health States, 4th ed., CM Porth, (2015), Wolters Kluwer, Philadelphia, USA.

3- Recommended Books:

i- Essentials of anatomy and physiology, 5th ed., VC Scanlo and T Sanders (2007), F. A. Davis Company, Philadelphia, UAS..

4- Periodicals and websites:

i- Advanced Physiology and Pathophysiology: Essentials for Clinical Practice https://www.springerpub.com/advanced-physiology-and-pathophysiology-9780826177070.html Course Coordinator: Assist. Prof. Dr. Waleed Barakat Head of Department: Prof. Dr. Mona Fouad

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	N	/latrix I	of Phys	siology d	& Patho	physiol	ogy			
		Key elements of Physiology & Pathophysiology course								
	Course Contents	1	-1- COMP	ETENCY	3-1- CON	MPETENC	ĊY		4-1- COMPETENCY	
	Lectures	1.C1.1	1.C1.2	1.C1.3	3.C1.1	3.C1.2	3.C1.3	3.C1.4	4.C	1.1
1	Cell physiology Neuromuscular and synaptic transmission	X	X	Х						
2	PNS Physiology Nervous system physiology	Х	X	Х						
3	Central nervous system physiology	X	X	Х						
4	Central nervous system physiology	X	X	Х						
5	Endocrine physiology	X	X	Х						
6	Renal physiology	X	X	Х						
7	Respiratory physiology	Х	Х	Х						
8	GIT physiology	Х	X	Х						
9	Cardiovascular system physiology	Х	Х	Х						
10	Cardiovascular system physiology	Х	Х	Х						
11	The lymphatic system and immunity	X	X	X						
12	The lymphatic system and immunity	X	X	Х						
13	Revision and open discussion	X	X	х						
	Practical session									
1	Lab safety rules				X	Х	Х	Х	Х	
2	Basic principles of case studies				Х	Х	Х	Х		
3	Pathophysiology of Psychosis Pathophysiology of Parkinsonism				Х	X	Х	Х	X	
4	Pathophysiology of Alzheimer disease Pathophysiology of Epilepsy				x	X	Х	Х		
5	Pathophysiology of Diabetes , other hormonal disorders				X	х	х	Х	X	

6	Pathophysiology of renal failure	X	Х	Х	Х		
7	Pathophysiology of Asthma Pathophysiology of Cough	X	Х	Х	Х	Х	
8	Pathophysiology of Hepatitis, Ulcer, Diarrhea, Constipation	X	Х	Х	Х		
9	Pathophysiology of Hypertension	Х	Х	Х	Х	Х	
10	Pathophysiology of Heart failure Pathophysiology of Angina	X	Х	Х	Х		
11	Pathophysiology of Inflammation	Х	Х	Х	Х	Х	
12	Pathophysiology Infectious disease	Х	Х	Х	Х		

Х

]	Matrix I	I of Physiolo	ogy & Pa	thoph	ysiology	v course	e			
	National Academic Program Course			Teach	ing and l methods	U	We	ighting o	f asses	sment		
Star	Academic Reference ndards NARS	Program key elements	key element s	Course contents	Sources		practical session	case study/ think- pair- share	written exam	practical exam	oral exam	Midterm exam
	Demonstrat e understanding of knowledge of pharmaceutical,			Cell physiology Neuromuscular and synaptic transmission PNS Physiology Nervous system physiology	Student book Essential books	x			x		x	x
1.1.1	biomedical, social, behavioral, administrative, and clinical sciences. Utilize the proper	1.C1.3 1.C1.8	1.C1.1 1.C1.2 1.C1.3	Central nervous system physiology Central nervous system physiology Endocrine physiology Renal physiology Respiratory physiology GIT physiology Cardiovascular system physiology	Student book Essential books	x			x		x	x
	pharmaceutical and medical terms, abbreviations and symbols in pharmacy practice			System physiology Cardiovascular system physiology The lymphatic system and immunity The lymphatic system and immunity	DOOKS							

				Lab safety rules Basic principles of case studies Pathophysiology of Psychosis	Practical notes	x		х	
3.1.1	Apply the principles of body function and basis of genomics in health and disease states to manage different diseases.	3.C1.1	3.C1.1 3.C1.2 3.C1.3 3.C1.4	Pathophysiology of Parkinsonism Pathophysiology of Alzheimer disease Pathophysiology of Epilepsy Pathophysiology of Diabetes , other hormonal disorders Pathophysiology of renal failure Pathophysiology of Asthma Pathophysiology of Cough Pathophysiology of Hepatitis, Ulcer, Diarrhea, Constipation Pathophysiology of Hypertension Pathophysiology of Heart failure Pathophysiology of Angina Pathophysiology of Angina Pathophysiology of Inflammation Pathophysiology of Inflammation Pathophysiology Infectious disease	Practical notes	X		X	
4.1.1	Demonstrate responsibility for team performance and peer evaluation of other team members, and	4.C1.1	4.C1.1	Activity			x		

express time						
management skills.						

Course Coordinator: Assist. Prof. Dr. Waleed Barakat

Head of Department: Prof. Dr. Mona Fouad

تم مناقشة و اعتماد توصيف المقرر من مجلس القسم بتاريخ / / Date: 2020

COURSE SPECIFICATIONS

General Microbiology and Immunology

Second Level– Semester 3 2020-2021

Course specification of General Microbiology and Immunology

University: Zagazig

Faculty: Pharmacy

A- Course specifications:

Program (s) on which the course is given: Bachelor of Pharmacy (Pharm D)Major or Minor element of programs:majorDepartment offering the program:------Department offering the course:Microbiology and ImmunologyAcademic year Level:Second Level/Semester3Date of specification approval:September 2020

B- Basic information:

Title: General Microbiology and Immunology Code: **PM 301** Credit Hours: 3 hrs Lectures: 2 hrs/week Practical: 1 hrs/week Tutorials: ---Total hours: 3 hrs/week

C- Professional information:

1-Overall aim of the course

On completion of the course, the student will have good knowledge about, microorganisms, their morphology, diversity, cell structure and function, cultural characteristics, growth, metabolism, role of microorganisms in infectious diseases and microbial pathogenesis. It also clarifies different mechanisms of transport across bacterial cell membrane, metabolic pathways and physiology of bacteria. The course also covers the principles of genetic characters including DNA and RNA structures, replication, different forms of mutation and mutagenic agents. It also explores the basic concepts microbial growth, cultivation and reproduction. Moreover it introduces the modern concepts of medical immunology, with an emphasis on Host parasite relationship, Non-specific and specific immunity, Mechanism of protective immunity. Molecular and cellular immunology, including antigen and antibody structure, function and reaction between them, effector mechanisms, complement, and cell mediated immunity. Active and passive immunization. Hypersensitivity and in vitro antigen antibody reactions, Immuno-deficiency disorders, Autoimmunity and auto-immune disease, organ transplantation.

2- 2- Key elements of General Microbiology and Immunology DOMAIN 1- FUNDAMENTAL KNOWLEDGE

1-1- COMPETENCY

Integrate knowledge from basic and applied pharmaceutical and clinical sciences to standardize materials, formulate and manufacture products, and deliver population and patient-centered care.

patient-ce	incred care.
	Illustrate different types of microorganisms and their way of life.
	Perform microscopical examinations, biochemical tests and
1.C1.1	serological reactions for identification of microorganisms
	Define the basic microbial growth conditions and metabolism.
1.C1.2	Monitor the microbial growth and growth conditions on different
	types of common culture media.
1.C1.3	Use the proper terms of microbiology and immunology.
	Illustrate the principles of immunology including natural and
1.C1.4	acquired immunity and antigen-antibody reactions. Identify the
	functions of immune system in health state and during disease state
1 C1 5	Outline the basis of bacterial genetics

1.C1.5 Outline the basis of bacterial genetics

DOMAIN 2: PROFESSIONAL AND ETHICAL PRACTICE

2-3- COMPETENCY

Handle and dispose biologicals and synthetic/natural pharmaceutical materials /products

effectively and safely with respect to relevant laws and legislations.

2.03.1	Handle basic laboratory equipments, chemicals and biohazards effectively and safely.
2.C3.2	Apply GLP guidelines for safe handling of chemicals and

biohazards

DOMAIN 3: PHARMACEUTICAL CARE

3-1- COMPETENCY

Apply the principles of body functions to participate in improving health care services using evidence-based data.

3.C1.1	Analyze and interpret experimental results of serological reactions
3.C1.2	Analyze and interpret experimental results for differentiation

DOMAIN 4: PERSONAL PRACTICE

4-1- COMPETENCY

Express leadership, time management, critical thinking, problem solving, independent and

team wor	rking, creativity and entrepreneurial skills.			
4.C1.1	4.C1.1 Work effectively as a member of a team			
4.C1.2	4.C1.2 Develop internet search and computer skills			

D- Contents:

Week	Lecture contents (2 hrs/week)	Practical session (1hrs/week)
1	 General introduction to microbiology and historical review Introduction to immunology 	 Laboratory safety measures Microscopy and general terms of microbiology
2	 Description of microorganisms Classification and types of Microorganisms Introduction to immunology 	• Microscopical examination of Bacteria: preparation and staining of smear, simple stain and negative stain
3	 Brief description of viruses, fungi and protozoa Immunity – innate immunity Immune system 	 Differential stains: Gram-stain
4	 Bacteria: description and classification Cells of immune response Immunogens or antigens 	 Differential stains: Gram-stain of mixtures of microorganisms Activity
5	 Anatomy and structure of bacterial cells Acquired immune response Cell mediated immunity 	 Differential stain: Acid-fast stain (Ziehl Neelsen stain) Examination of living bacteria: hanging drop technique
6	 Growth and cultivation of bacteria, bacterial growth curve Humoral immune response And Cytokines 	 Spore stain Microscopic examination of fungi: lactophenol mount
7	Mid-ter	m exam
8	 Microbial metabolism Agglutination and complement fixation reactions 	• Cultivation of bacteria: types of common culture media and growth conditions
9	 Microbial metabolism Immunologic mechanisms of tissue damage 	• Biochemical activities of and identification of bacteria
10	Microbial metabolismHypersensitivity reactions	 Serological reactions (Precipitation reactions) Activity
11	Microbial geneticsTransplantation immunology	• Serological reactions (Agglutination reactions)
12	• Transcription and Protein synthesis	Serological reactions

	Autoimmune diseases	(Complement fixation reaction
13	Genetic variation	Final practical exam
	Tumour immunology	
14	Genetic Transfer among bacteria	
	 Immunoprophylaxis 	
15	• Written exam	

E- Teaching and Learning Methods:

- Lectures
- Practical sessions
- Internet search and poster preparation
- Others : videos
- Blended learning

F- Student Assessment methods:

- 1- Written exams to assess: 1.C1.1, 1.C1.2, 1.C1.3, 1.C1.4, 1.C1.5, 3.C1.1, 3.C1.2
- 2- Activity to assess: 4.C1.1, 4.C1.2
- 3- Practical exams to assess: 2.C3.1, 2.C3.2, 3.C1.1, 3.C1.2
- 4- Oral exam to assess1.C1.1, 1.C1.2, 1.C1.3, 1.C1.4, 1.C1.5, 3.C1.1, 3.C1.2

Assessment schedule

Assesment (1): Midterm exam	Week 7
Assessment (2): Activity (Poster)	Week 4, 10
Assessment (3): Practical exams	Week 13
Assessment (4): Final written exam	Week 15
Assessment (5): Oral exams	Week 15

Weighting of Assessment

Assessment method	Marks	Percentage
Activity & Periodical exam	15	15%
Practical exam	25	25%
Final written exam	50	50%

Oral exam	10	10%
TOTAL	100	100%

G- Facilities required for teaching and learning:

- 1. For lectures: Black (white) boards, and data show.
- 2. For Labs.: Chemicals, Autoclaves, Incubators, Ovens, Water bathes, staining

dyes, microscopes, refrigerators and microbiological culture media

H- List of References:

1- Course Notes: Student book of General Microbiology and Immunology Approved by Microbiology and Immunology department

2- Essential Books:

- Jackson M, Lowey A. Handbook of extemporaneous preparation. A guide to pharmaceutical compounding. Published by Pharmaceutical Press, 2010.

3- Recommended Books

- Martindale, "The extra pharmacopeia". 31st edn, by James, E.F Reynolds. And Kathleen Parfitt, Royal Pharmaceutical Society, London (2007).

4- Periodicals and websites:

 Aquilina A. The extemporaneous compounding of paediatric medicines at Mater Dei Hospital. Journal of the Malta College of Pharmacy Practice.Issue 19, 28 – 30, 2013. <u>http://canadianpharmacistsletter.therapeuticresearch.com/ce/ceCourse.asp...</u>

Course Coordinator: Prof Dr/ Nehal Elsayed

Head of Department: Prof / Nehal Elsayed Yousef

تم مناقشة و اعتماد توصيف المقرر من مجلس القسم بتاريخ Date

				Matr	ix1 of	f Gen	<mark>eral</mark> I	Aicro	biolo	gy and I	<mark>lmmur</mark>	nology	
					Key elements of principles of General Microbiology and Immunology						y		
Co	ourse conte	nt	DC		- FUND. OWLED	AMENT. GE	AL	DOMAIN 2: PROFESSION		DOMA PHARMA AL C	CEUTIC	PERS	IAIN 4: SONAL CTICE
		Γ	1.C1.1	1.C1.2	1.C1.3	1.C1.4	1.C1.5	2.C3.1	2.C3.2	3.C1.1	3.C1.2	4.C1.1	4.C1.2
1	 General introducti microbiol and histor review Introducti immunolo <u>Practical</u> Microscoj general te microbiol 	ogy ical on to ogy py & rms of ogy	x		x	x		,	X				
2	 Description microorga and types Microorga Introducti immunolo Practical Microscop examinati Bacteria b simple an negative s 	anisms of anisms on to ogy pical on of oy d	x			x		2	ζ.		X		
3	 Brief descriptio viruses, fu and proto Immunity innate imm Immune s <u>Practical</u> Differenti 	n of ingi zoa — munity ystem	x			x		2	ζ.	X	X		

	stains: Gram-stain								
4	 Bacteria: description and classification Cells of immune response Immunogens or antigens <u>Practical</u> : Gram-stain: mixture Activity 		X	X			X	x	x
5	 Anatomy & structure of bacterial cells Acquired immune response: Cell mediated immunity Practical Differential stain: (Acid-fast stain) Examination of living bacteria: 	X		X					
6	 Growth and cultivation of bacteria, bacterial growth curve Humoral immune response and Cytokines <u>Practical</u> Spore stain 		X		X	X			

	 Microscopic 							
	examination of							
	fungi							
7	• Midterm exam				X			
8	Microbial metabolism Agglutination and CFT <u>Practical</u> Cultivation of bacteria: types of common culture media and growth conditions	x				x		
9	Microbial metabolism Immunologic mechanisms of tissue damage <u>Practical</u> Biochemical activities of and identification of bacteria	X	х					
10	Microbial metabolism • Hypersensitivity reactions <u>Practical</u> • serological reactions <u>Activity</u>	x	x		x		x	x
11	 Microbial genetics Transplantation immunology <u>Practical</u> serological reactions 	x	x		x			
12	 Transcription and Protein synthesis Autoimmune diseases <u>Practical</u> serological reactions 		X	x	X			
13	 Genetic variation 		Х	X	Х			

Tum	nour immunology						
14 a	enetic Transfer mong bacteria nunoprophylaxis		X	x	X		

			Matrix2	of General M	licrobio	logy ar	nd Imm	unolog	y			
		Program	Course	C		Teach	ing and l method	learning s	M	ethod of	assess	ment
	National Academic Reference Standards (NARS) Principles of basic, pharmaceutical, medical, social, behavioral, management, health and environmental sciences as well as pharmacy practice	key elements	key elements	Course contents	Sources	lecture	practical session	Activity	written exam	practical exam	oral exam	Midterm exam
		1.C1.2	1.C1.2	•Growth and cultivation of bacteria, bacterial growth curve	Student book Essential books	x			Х		X	х
		1.01.2	1.C1.5	•Microbial genetics •Genetic variation •Genetic Transfer among bacteria	Student book Essential books	x			X		X	Х
1-1-1	pharmaceutical, medical, social, behavioral, management, health and environmental sciences as well as pharmacy		1.C1.3	 Introduction to immunology Acquired immune response 1.Cell mediated immunity Humoral immune response And Cytokines 	Student book Essential books	x			Х		X	Х
	practice	1.C1.3	1.C1.4	Cells of immune response Immunogens or antigens •Agglutination and complement fixation reactions •Hypersensitivity reactions	Student book Essential books	x			X		X	X

2.3.1Handle and dispose chemicals and pharmaceutical preparations safely.2.C3.1.Ceneral introduction to microbiology and historical review to immunologyxx2.3.2Recognize and adopt ethical, legal, and safety guidelines for handling and disposal of biologicals, and pharmaceutical materials/products.2.C3.2	1.1.2	Use the proper pharmaceutical and medical terms, abbreviations and symbols in pharmacy practice.	1.C1.8	1.C1.3	•General introduction to microbiology and historical review •Introduction to immunology		x		x	x	x
diseases.	2.3.2	chemicals and pharmaceutical preparations safely. Recognize and adopt ethical, legal, and safety guidelines for handling and disposal of biologicals, and pharmaceutical materials/products. Monitor and control microbial growth and carry out laboratory tests for identification of infections/	2.C3.2	2.C3.2 3.C1.1	 introduction to microbiology and historical review Introduction to immunology Agglutination and complement fixation reactions Growth and cultivation of bacteria, bacterial 	Practical notes practical					

4.1.2 Retrieve and evaluate information from different sources to improve Professional competencies.4.C1.44.C1.2Practical notes Recommen ded books Internetxx	4-1-1	Manage time as evidenced by the ability to plan and implement efficient mode of working.	4.C1.3	4.C1.1	Student book practical notes	x		x	x
	4.1.2	evaluate information from different sources to improve	4.C1.4	4.C1.2	notes Recommen ded books	x			x

COURSE SPECIFICATIONS

Pharmaceutics I

Second Level– Semester 3 2020-2021

Course specification of Pharmaceutics I

University:	Zagazig	Faculty:	Pharmacy
A- Course s	pecifications:		
Program (s) or	n which the course is g	given: Bachelor of Phar	macy (Pharm D)
Major or Minc	or element of program	s: Major	
Department of	fering the program:		
Department of	fering the course:	Pharmaceutics Depar	tment
Academic year	r Level:	Second level/ semes	ter3
Date of specifi	cation approval:	September 2020)
B- Basic inf	ormation:		

Code: **PT 303**

Title: Pharmaceutics I Credit Hours: ---Lectures: 2hr/week Practical: 1hr/week Tutorials: ---Total: 3 hrs/week

C- Professional information:

1-Overall aim of the course

This course provides students with knowledge of the system of weights, measures, mathematical expertise and pharmaceutical calculations requisite to the compounding, dispensing, and utilization of drugs in pharmacy practice. It is also concerned with all manufacturing formulations aspects, packaging, storage and stability of liquid dosage forms including solutions (aqueous and non-aqueous), suspensions, emulsions and colloids with emphasis on the technology and pharmaceutical rationale fundamental to their design and development. The incompatibilities occurring during dispensing are also considered.

2- Key elements of Pharmaceutics I

DOMAIN	N 1- FUNDAMENTAL KNOWLEDGE
pharmace	APETENCY: Integrate knowledge from basic and applied utical and clinical sciences to standardize materials, formulate and ure products, and deliver population and patient-centered care.
1.C1.1.	Define different liquid dosage forms, different measuring systems and different types of
	incompatibility.
	Enumerate the ideal characters of different liquid
1.C1.2.	pharmaceutical dosage forms; solutions, emulsions, suspensions, colloids.
1 01 2	Describe the different methods for preparation of
1.C1.3.	solutions, emulsions, suspensions and colloids.
1 01 4	Illustrate the ideal characters for packaging, labeling,
1.C1.4.	storing and distribution process in industry
1.C1.5	Select the appropriate ingredients used in formulation of
1.01.5	different liquid dosage forms
DOMAIN	N 2: PROFESSIONAL AND ETHICAL PRACTICE
manufact	IPETENCY: Standardize pharmaceutical materials, formulate and ure pharmaceutical products, and participate in systems for g, storage, and distribution of medicines.
2.C2.1	Apply good handling and disposal skills of different
2.02.1	chemicals and pharmaceutical preparations.
2.C2.2.	Perform different pharmaceutical calculations
	Compound, dispense and label different pharmaceutical
2.C2 3.	dosage forms safely and effectively
2.C2.4	Identify incompatibility problems with proper
2.02.4	overcoming strategies
	IPETENCYL: Handle and dispose biologicals and
-	natural pharmaceutical materials /products effectively and safely ect to relevant laws and legislations.
2.C3.1.	Handle pharmaceutical preparations safely
	A DEDCONAL DDACTICE
DOMAIN	N 4: PERSONAL PRACTICE

Express leadership, time management, critical thinking, problem solving,	
independent and team working, creativity and entrepreneurial skills.	
1 C1 1 Develop team working skills	

4.C1.1. Develop team working skills

D- Contents:

Week	Lecture contents (2 hr/week)	Practical session (1hr/week)
No.		
1	Fundamentals of measurements and	
1	calculations	
	- Arabic and Roman Numerals	Pharmaceutical calculations:
	Systems of Measurement	Measurement Systems
	-The Metric System	
2	Systems of Measurement (Cont.)	Pharmaceutical calculations:
	- Apothecary and Avoirdupois systems	Dilution and concentration of
	(The Common Systems)	Solutions
3	Household systemDilution and concentration of	
5	Solutions	Pharmaceutical calculations:
	- Reducing and Enlarging Formulas,	Reducing and Enlarging Formulas,
	- percentage and ratio strength	percentage and ratio strength
	problems	problems
	- Calculation of Doses for Children	
4	- Introduction to liquid dosage forms	
	Solutions: Advantages and	-Preparation of aromatic water
	disadvantages Of Solutions, Aqueous	-Preparation of Simple mixture of
	liquid dosage forms pharmaceutical Solutions	liquorice
5	Solutions: Non aqueous liquid dosage	-Preparation Ammonium Chloride
	forms Sweet and/or viscid liquid dosage	Cough Mixture
	forms, Manufacture of solutions	-Preparation of Ear drops
6	Emulsion: General description,	-Methods of preparation of
	Advantages and disadvantages of	emulsions
	pharmaceutical emulsions, emulsion instability and theories of emulsification	a- wet method b- dry method
7	Mid-term exam	
8	Emulsion: Formulation of pharmaceutical emulsions,	
	Excipients used in pharmaceutical	Castor oil emulsion (wet method)
	emulsions,	Castor oil emulsion (Dry Method)
	Manufacture of emulsions	
9	Suspension: Introduction, advantages	Minoral oil amulaion (wat mathed)
	and disadvantages of Suspension,	Mineral oil emulsion (wet method) Mineral oil emulsion (Dry
	Reasons for preparing suspension,	Method)
10	Characters of ideal suspension	
10	Suspension: Formulation and 104	preparation of Suspension

	evaluation of suspensions, Stability of suspensions	
11	Colloids: introduction, Pharmaceutical application of colloids, Types of colloidal systems	determination of sedimentation rate
12	Colloids: Preparation of colloids, purification, Properties of colloids (kinetic, optical, electrical)	Incompatibility
13	-Stability of colloids	Practical exam
	- Electrical Double Layer, Nernst And Zeta Potentials, Importance Of Zeta Potential	
14	Zeta Potentials, Importance Of Zeta	

E- Teaching and Learning Methods:

- Lectures
- Practical session
- Think/pair/share
- Blended learning

F- Student Assessment methods:

- 1- Written exams to assess: 1.C1.1, 1.C1.2, 1.C1.3, 1.C1.4, 1.C1.5
- 2- Practical exams to assess: 2.C2.1, 2.C2.2, 2.C2.3, 2.C2.4, 2.C3.1
- 3- Activity within labs: 4.C1.1
- 4- Oral exam to assess: 1.C1.1, 1.C1.2, 1.C1.3, 1.C1.4, 1.C1.5

Assessment schedule

Assessment (1): Periodical exams	Week 7
Assessment (2): Final written exams	Week 15
Assessment (3):Practical exams & practical work	Weekly
Assessment (4):Oral exams	Week 15

Weighting of Assessment

Assessment method	Marks	Percentage
Periodical exam	10	10%

Final Written exam	50	50%
Activities within labs	5	5%
Practical exams	25	25%
Oral exam	10	10%
TOTAL	100	100%

G- Facilities required for teaching and learning:

For lectures: Black (white) boards, data show

For labs: Chemicals, glass ware, digital balance, water bathes

H- List of References:

1- Course Notes: Student book of physical pharmacy approved by pharmaceutics department

2020-2021

2- Essential Books:

- The science of dosage form design, Aulton, M. E., 2nd edition, Churchill Livingstone, London. (2002).
- Pharmaceutical Dosage Forms: Rational design and formulation with excipients, Larry L. Augsburger, Stephen W. Hoag, Informa Healthcare USA, (2008)
- Pharmaceutical Calculations, 12th edition, Howard C. Ansel and Mitchell J. Stoklosa.(2005)
- Aulton, Michael E. Pharmaceutics: The Science of Dosage Form Design. 2 nded.Churchill, 2002

3- Recommended Books:

- Remington's Pharmaceutical Science. Alfonso, Gennaro, R., ^{17 th} edn, Mack
 Publishing Company, USA. (1985).
- Handbook of Pharmaceutical Manufacturing Formulations: Liquid products, <u>Sarfaraz Niazi</u>, Sarfaraz K. Niazi, CRC Press, (2004).
- Pharmacy Calculations for Technicians, 3rd edition, EMCParadigm puplishing.
 Don A. Ballington and Tova Wiegand Green. (2007)

4- Periodicals and websites:

www.emedicine.com www.sciencedirect.com www.pubmed.com Course Coordinator: Dr. Azza Ali Hasan Soliman

Head of Department: Dr. Nagia Ahmed El-megrab

Date:

تم مناقشة و اعتماد توصيف المقرر من مجلس القسم بتاريخ / /2020 م

				Ma	trix I	of Ph	arma						
								Key	elemo	ents of	f Pharn	naceutics I cour	se
	Course Contents Lectures			1-1- COMPETENCY				2-2	- COM	PETEN	NCY	2-3- COMPETENC Y	4-1- COMPETENCY
				1.C1 .2	1.C1 .3	1.C1 .4	1.C1 .5	2.C2 .1	2.C2 .2	2.C2 .3	2.C2. 4	2.C3.1	4.C1.1
1	Systems of	Measurement	X										
2	Systems of	Measurement (Cont.)	X										
3	Different C	alculations problems	X										
4	Liquid dosa Solutions	age forms	X	X	X	X	X						
5	Solutions (Cont.)	Х			х	х						
6	Emulsion		X	х	x	х	х						
7	Emulsion (Х	x	X	х	Х						
8	Suspension		X	x	x	X	Х						
9	Suspension	(Cont.)	X	Х		Х	Х						
10	Colloids		X	x	x	х	Х						
11	Colloids (C		x	X	x	x	x						
12		Double Layer, Nernst And Zeta Importance Of Zeta Potential		X									
13	Incompatib		x								Х		
		Practical session											
1	Measuremen	nt Systems						X	X	X			

2	2 Dilution and concentration of Solutions		v	v	х			
3	 Reducing and Enlarging Formulas, percentage and ratio strength problems 		x	x	X			
4	 4 -Preparation of aromatic water -Preparation of Simple mixture of liquorice 		x	x	x		X	X
5	 Preparation Ammonium Chloride Cough Mixture Preparation of Ear drops Image: Ammonium Chloride Cough Image: Ammonium Chlored Cough Image: Ammonium Chloride Cough		X	x	х		х	
6	• -Methods of preparation of emulsions		x	X	х		X	
7	 Castor oil emulsion (wet method) Castor oil emulsion (Dry Method) 		x	X	x		x	
8	8 Mineral oil emulsion (wet method) Mineral oil emulsion (Dry Method)		x	X	х		X	X
9	9 preparation of Suspension		x	X	x		x	
1 0	1 determination of sedimentation rate		x	x	X		x	
1 1	1 1Incompatibility					X		x

			N	Aatrix II of P	harmac	eutics	I course	9				
	National Academic		Course key elements		Sources	Teaching and learning methods			Weighting of assessment			
	Academic Reference ndards NARS	e Program key elements		Course contents		lecture	practical session	case study/ think- pair- share self learning	written exam	practical exam& activity	oral exam	Midterm exam
1.1.1	Demonstrate understanding of knowledge of pharmaceutical, biomedical, social, behavioral, administrative, and clinical sciences.	1.C1.2	1.C1.1	Systems of Measurement Systems of Measurement (Cont.) Different Calculations problems Liquid dosage forms Solutions Solutions Solutions Solutions (Cont.) Emulsion Emulsion (Cont.) Suspension Suspension (Cont.)	Student book Essential books	x			x		x	x

			1.C1.2 1.C1.3 1.C1.4	Liquid dosage forms Solutions Solutions (Cont.) Emulsion Emulsion (Cont.) Suspension Suspension (Cont.) Colloids	Student book Essential books Student books Student book Essential books Student	x x x		x x x		x x x	x x x
			1.C1.5	Colloids (Cont.)	book Essential books	x		х		x	x
2-2-22	Apply the basic requirements of quality management system in developing, manufacturing, analyzing, storing, and distributing pharmaceutical materials/ products considering various incompatibilities.	2.C2.2. 2.C2.3	2.C2.1 2.C2.2 2.C2.3 2.C2.4	Measurement Systems Dilution and concentration of Solutions Reducing and Enlarging Formulas, percentage and ratio strength problems -Preparation of aromatic water -Preparation of	Practical notes		X		x		

2.3.1	Handle, identify, and dispose biologicals, synthetic/natural materials, biotechnology-based and radio-labeled products, and other materials/products used in pharmaceutical field.	2.C3.1	2.C3.1	Simple mixture of liquorice -Preparation Ammonium Chloride Cough Mixture -Preparation of Ear drops -Methods of preparation of emulsions Castor oil emulsion (wet method) Castor oil emulsion (Dry Method) Mineral oil emulsion (wet method) Mineral oil emulsion (Dry		x		x		
				Mineral oil						
				preparation of						
				Suspension						
				determination of						
				sedimentation rate						

			Incompatibility					
Demonstrate responsibility for team performance and peer evaluation of other team members, and express time management skills.	4.C1.1	4.C1.1		Practical notes	x		x	

Course Coordinator: Dr. Azza Ali Hasan Soliman

Head of Department: Dr. Nagia Ahmed El Megrab

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